Republic of Iraq The Ministry of Higher Education & Scientific Research



University:Al Muthanna College: Science Department: Chemistry Stage: First Lecturer name: Azal Shaker Academic Status: Assistant professor Qualification: Ph.D Place of work: Chemistry dep.

Course Weekly Outline

Course Instructor		Azal Shaker					
Email	azilshker@mu.edu.iq						
Title	Inorganic chemistry						
Course Coordinator	The first course						
Course Objective	Students will understand the development of the structure of atom, evidence for various atomic model, electron energy levels, atomic spectra, quantum numbers, periodic table and periodic Law.						
Course Description	This is theoretical course designed to describe and explain the historical development of atomic structure . it reviews various model about the structure of atom and how these models related to some aspects of atom , e.g. atomic spectra . the course further provide understanding of the dual property of the electron and how this leads to the quantum mechanical model of atomic structure , Furthermore , the pattern of electron distribution in atom will be explained as well as how to specify a particular electron in an atom						
Textbook	Inorganic chemistry Thanaa al hassany						
References	Inorganic chemistry(Huhhey) Shriver and Atkins Inorganic chemistry, 5th ed., 2010						
	Term Tests	Laboratory	Quizzes	Project	Final Exam		
Course Assessment	As (40%)	(0%)	(0%)		As (60%)		
General Notes							

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Course weekly Outline

week	Date	Topics Covered	Lab. Experiment	Notes
			Assignments	
1 2022/11/2	2022/11/2	Atomic structure, Dalton, Thomson		
	, and Rutherford atomic models			
2		Quantum theory, Electromagnetic		
2022/11/9	radiation, Photoelectric effect, Black			
		body		
3		Atomic spectra and electromagnetic		
2022/11/16	spectrum, Emission spectra,			
	2022/11/16	Absorption spectra, Hydrogen atom		
		spectra		
4		Bohr atomic model , Application		
2022/11/23		Bohr theory on atomic similar		
		Hydrogen		
5		Somerfield theory, Zeeman effect		
2022/11		effect of spin electron . Quantum		
	2022/11/30	number , physical description of		
		atomic orbitals		
6 2022/12/7		Wave mechanics. Uncertainty		
	2022/12/7	principle. Schrodinger Equation.		
		Similar orbitals, nonSimilar orbitals		
7		The Arrangement of electrons in		
	2022/12/14	atoms Pauli exclusion principle		
		Hund first rule		
8		Periodic Table, typical element, the		
	2022/12/21	representative elements the main		
		transition elements		
9		Russel –Sounders symbols (Term		+
	2022/12/28	symbols) examples Shielding		
	2022/12/20	Jonization anargy		
10	2022/1//	Fffootivo nucleor chorge		<u> </u>
10	2023/1/4	Electron offinity cleatron and it.		
11	2023/1/11	Liectron attinity, electronegativity		

Instructor Signature: Azal shaker **Dean Signature:**