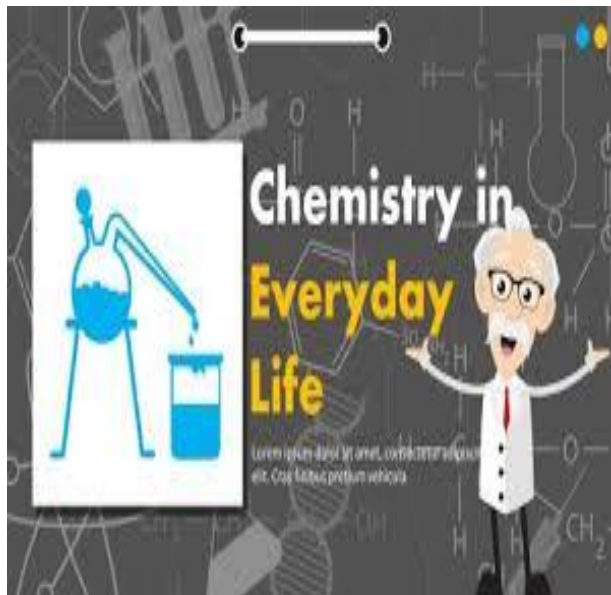


Fundamentals of Analytical Chemistry

م.د. مسار علي عواد



What is Analytical Chemistry?



Analytical chemistry is a measurement science consisting of a set of powerful ideas and methods that are useful in all fields of science and medicine.

“ **Analytical chemistry** is a science of chemical characterization and measurement.”

H. A. Laitinen, 1982

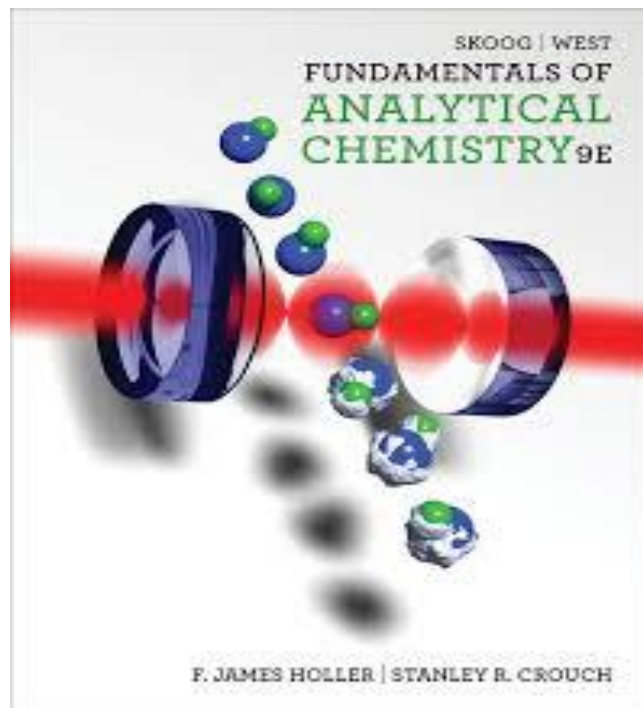
Principle of classical method;

1) **stoichiometry**

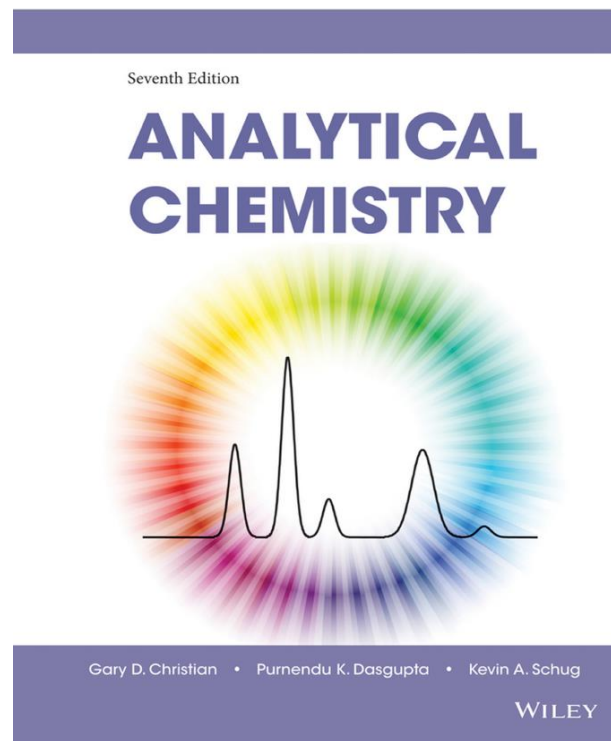
2) **equilibrium state**

References

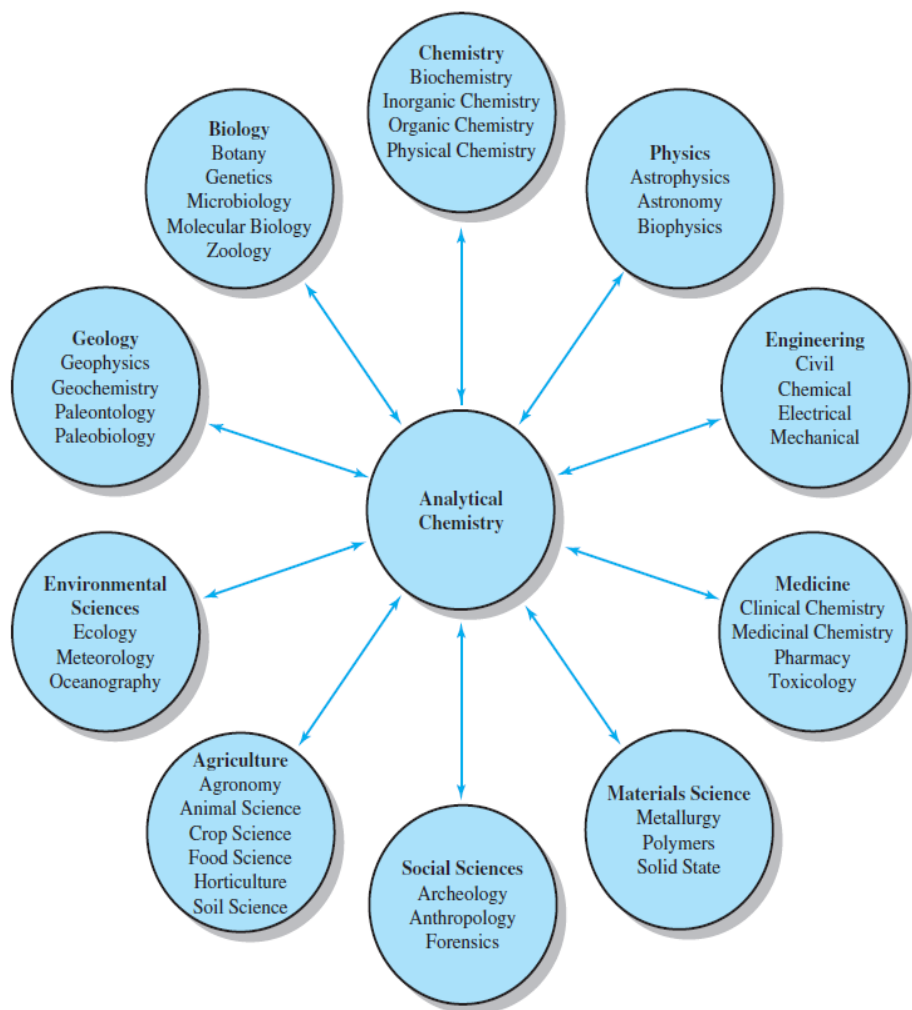
**Fundamentals of Analytical Chemistry,
Skoog, Holler, and West, Brooks/Cole**



**Analytical Chemistry, 2004,
Gary D. Christian, 6th Ed**

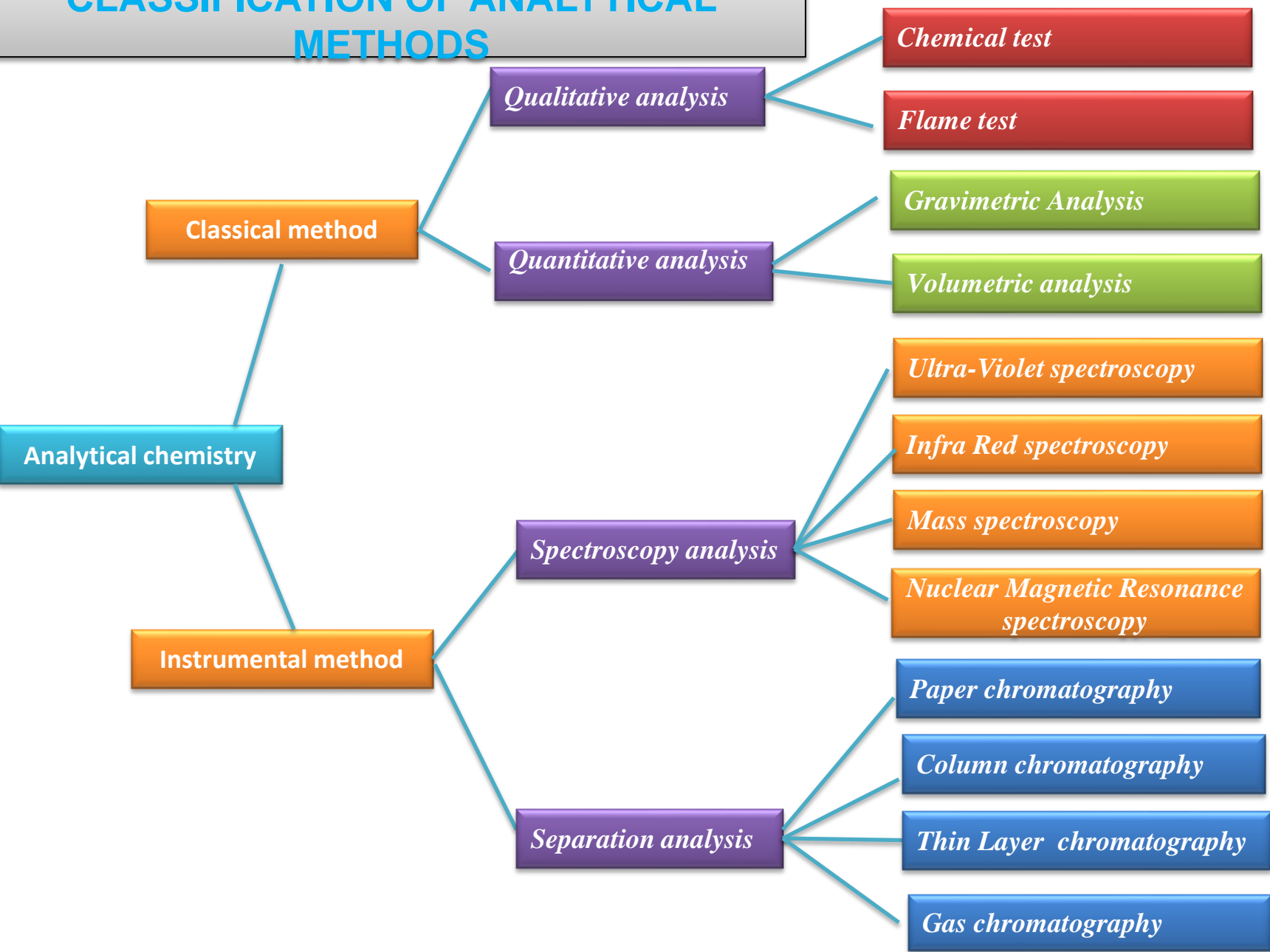


The relationship between analytical chemistry, other branches of chemistry, and the other sciences

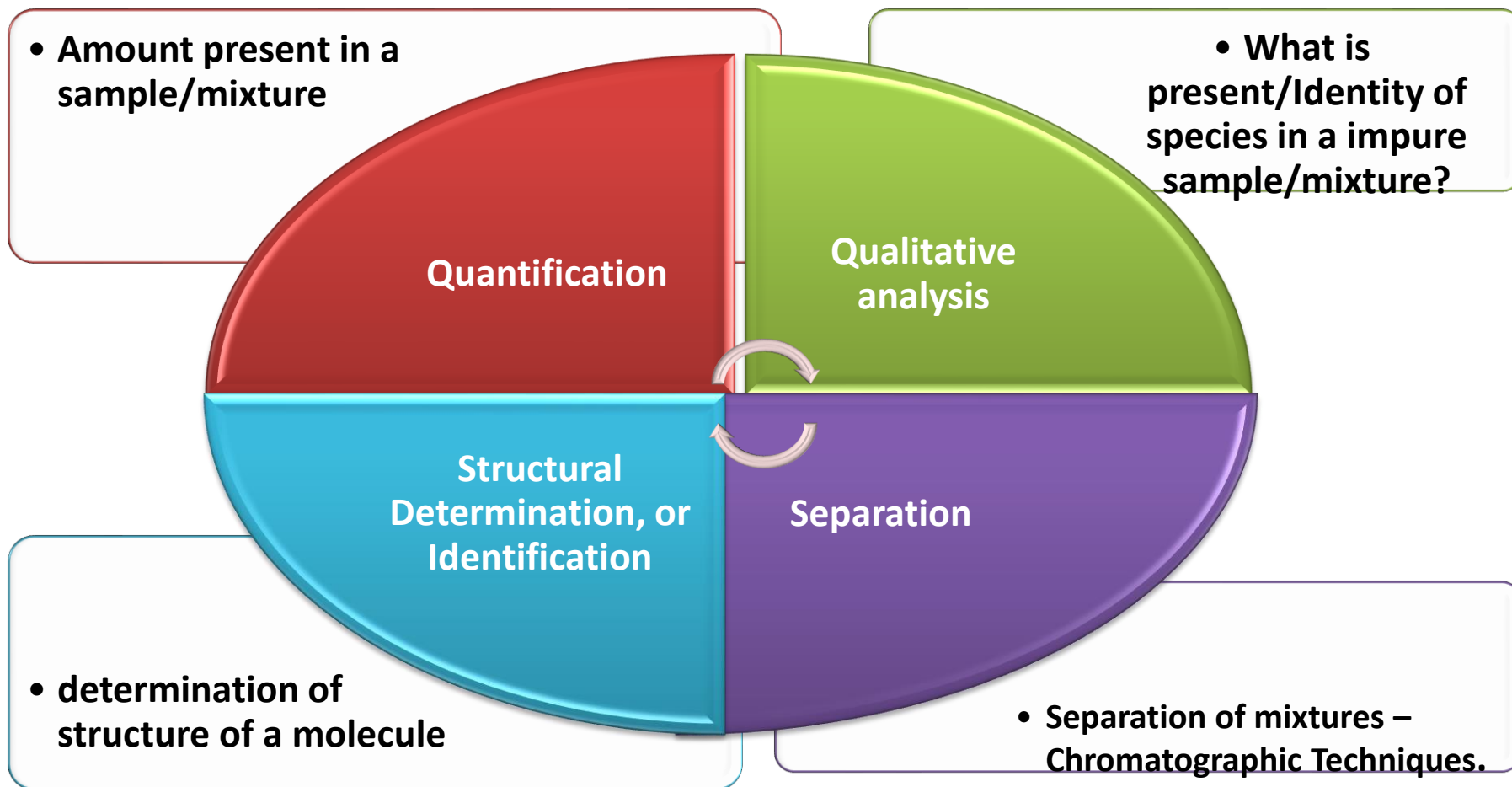


- **Clinical analysis** - blood, urine, feces, cellular fluids, etc.,
- **Pharmaceutical analysis** - establish the physical properties, toxicity, metabolites, quality control, etc.
- **Environmental analysis** - pollutants, soil and water analysis, pesticides.
- **Forensic analysis** - analysis related to criminology; DNA finger printing, finger print detection; blood analysis.
- **Industrial quality control** - required by most companies to control product quality.
- **Bioanalytical chemistry and analysis** - detection and/or analysis of biological components (i.e., proteins, DNA, RNA, carbohydrates, metabolites, etc.).
- **Archeology, Geology, Food Analysis,**

CLASSIFICATION OF ANALYTICAL METHODS



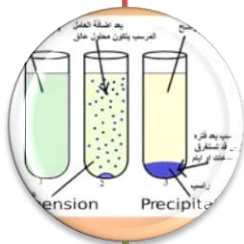
Analytical Chemistry deals with methods for



CLASSIFICATION OF ANALYTICAL METHODS



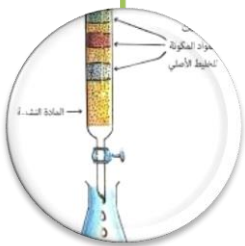
Analytical
Techniques



Chemical Methods
(Classical Method)



Physical Methods
(Instrumental
Method)



Quantitative



Qualitative



Separation

Instrumental methods

Qualitative – chromatography, electrophoresis and identification by measuring physical property.

(e.g. spectroscopy, electrode potential)

Quantitative – measuring property or determining concentration (e.g. Spectrophotometry, mass spectrometry)

Often, same instrumental method used for qualitative and quantitative analysis.



Classical methods

Qualitative

identification by color, indicators, boiling or melting points, odors

used to identify and separate cations and anions in a sample substance

Quantitative

mass or volume (e.g. gravimetric, volumetric

-